CHROM. 5 II5

# SEPARATION AND IDENTIFICATION OF FOOD COLOURS 

# I. IDENTIFICATION OF SYNTHETIC WATER SOLUBLE FOOD COLOURS USING THIN-LAYER CHROMATOGRAPHY 

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## SUMMARY

A thin-layer chromatographic method is described for the separation and identification of forty-nine synthetic food colours which are used in food products or which have been used. $R_{F}$ and $R_{X}$ (with respect to Orange G) values are tabulated and a scheme for the rapid iclentification of the components of a mixture of dyes is proposed.

## INTRODUCTION

## Colouring matters in food

Processed foods are often coloured to retain the appearance of the original material and to provide a more appealing product. Foodstuffs may be coloured by (a) synthetic organic dyestuffs, (b) inorganic pigments and (c) natural colouring materials obtained from vegetable and animal sources. Synthetic organic dyestuffs are generally used. However, no two countries in the world have identical lists of permitted food colours because there are differences of opinion about the toxicity of the various food colours. Consequently it is possible that foodstuffs may be imported into a country which forbids the colouring matters present in the products. A method has been developed for the identification of synthetic food colours using thin-layer chromatography. The dyes covered by the method are those which are permitted in countries, who are members of the Codex Alimentarius Commission, or dyes which have been used in the past but are now considered too harmful for use in foodstuffs. These dyes are listed in Table I together with their colour index number and the countries in which they are permitted.

A number of thin-layer chromatographic separations of water soluble dyes used in food have been described but most of these deal with only a limited number of dyes; normally those permitted in one country only or a group of dyes of similar

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TABLE I
FOOD DYES PERMITTED IN VARIOUS COUNTRIES




TABLIE II
CODES FOR DYES CHROMATOGRAPHED IN SOLVENTS $1,2,3, A N D$ 4

| Code | Possible identity of dye | Code | Possible identity of dye | Code | Possible identity of dye |
| :---: | :---: | :---: | :---: | :---: | :---: |
| AAAA | Blue VRS | DADC | Orange RN | EBDE | Carmoisine |
|  | Brilliant Blue | DBCA | Naphthol Yellow S | EBEB | Quinoline Yellow |
|  | Light Green | DBCB | Orange GGN | EBEC | Quinoline Yellow |
|  | Yellowish |  | Sunset Yellow | ECCA | Orange IRN |
|  | Patent Blue V | DBCC | Orange GGN | ECCB | Fast Red E |
| AAAB | Brilliant Blue |  | Sunset Yellow | ECCC | Fast Red E |
|  | Fast Green | DCCA | Orange RN |  | Incligo Carmine |
|  | Green S | DCCB | Orange GGN |  | Orange IRN |
|  | Light Green |  | Sunset Yellow |  | Red Io $B$ |
|  | Yellowish | DCCC | Acid Yellow | ECCD | Carmoisine |
|  | Patent Blue V |  | Orange GGN |  | Indigo Carmine |
|  | Yellow 2G |  | Orange RN |  | Red Io B |
| AAAC | Yellow 2G |  | Sunset Yellow | ECCE | Carmoisine |
| $A A B A$ | Scarlet GN | DCCD | Acid Yellow |  | Red io B |
| AACA | Scarlet GN |  | Orange GGN | ECDA | Orange RNN |
| ABAA | Brilliant Blue |  | Sunset Yellow | ECDB | Fast Red E |
|  | Light Green | DCDA | Orange RN | ECDC | Bordeaux B |
|  | Yellowish | DCDC | Orange RN |  | Fast Red E |
| ABAB | Brilliant Blue | DCDD | Red 6B |  | Indigo Carmine |
|  | Light Green | DCDE | Red 6B |  | Orange RN |
|  | Yellowish | DCED | Red 6B |  | Ponceau 3R |
| ABAC | Acid Magenta | DCEE | Red 6B |  | Ponceau MX |
| ABAD | Acid Magenta | DDDD | Amaranth |  | Ponceau SX |
| $A C A C$ | Acid Magenta |  | Red 6B |  | Red ro B |
| ACAD | Acid Magenta | DDDE | Red 6B | ECDD | Carmoisine |
| ACCD | Tartrazine | DDED | Red 6B |  | Indigo Carmine |
| ADCD | Ponceau 6R | DDEE | Red 6B |  | Ponceau SX |
|  | 'rartrazine | DEDD | Red 6B |  | Red 6B |
| ADCE | Ponceau 6R | DEDE | Red 6B |  | Red rob |
| AECD | Ponceau 6R | DEED | Red 6B | ECDE | Carmoisine |
| AECE | Ponceau 6R | DEEE | Red 6B |  | Red 6B |
| BAAA |  | EAAA |  | IECEC |  |
|  | Violet 5BN |  | Methyl Violet |  | Bordcaux 13 |
|  | Violet BNP |  | Rhodamine B |  | Ponceau 3 R |
| BBBB | Orange G |  | Violet 6B |  | Ponceau MX |
| BCBC | Ponceau $4 R$ | EABA | Auramine |  | Ponceau SX |
| BCCC | Ponceau 4 R |  | Methyl Violet | ECED | Ponccau SX |
| BCCD | Tartrazine |  | Fhodamine B |  | Red 6B |
| BDCD | Tartrazine | EACA | Auramine | ECEE | Red 6B |
| CAAA | Guinea Green B |  | Eosine | EDCC | Indigo Carmine |
|  | Violet BNP |  | Erythrosine |  | Red Io B |
|  | Violet 5BN |  | Chrysoidine | EDCD | Indigo Carmine |
|  | Violet 6B |  | Orange I |  | Red ro B |
| CACB | Chrysoin S |  | Orange RN | EDCE | Red ro B |
| CACC | Chrysoin S | EACB | Chrysoidine | EDDC | Indigo Carmine |
| CBCA | Naphthol Yellow S |  | Eosine |  | Red io B |
| CBCB | Orange GGN |  | Erythrosine | EDDD | Indigo Carmine |
|  | Sunset Yellow |  | Orange I |  | Red 6B |
| CBCC | Orange GGN | EACC | Chrysoin 5 |  | Red ro B |
|  | Sunset Yellow |  | Orange RN | EDDE | Red 6B |
| $\begin{aligned} & \text { CCBC } \\ & \text { CCCB } \end{aligned}$ | Ponceau 4 R Acid Yellow | EADA | Chrysoidine Orange I | EDED | Red ro B ${ }^{\text {Black } 7984}$ |
|  | Orange GGN |  | Orange RN |  | Black PN |
|  | Sunset Yellow | EADB | Orange I |  | Red 6B |
| CCCC | Acid Yellow Orange GGN | EADC | Quinoline Yellow Orange RN | EDEE | Black 7984 Black PN |

TABLEII (continued)

colour ${ }^{1-13}$. Cellulose and silica gel appeared to be the two most promising adsorbents for the separation of the water soluble dyes and so we have used only these two adsorbents with a variety of development solvents as listed in Table III.

A scheme for the quick identification of a colour or mixture of colours is proposed which is not dependent on the measurement of $R_{F}$ values. This consists of running the dye or mixture of dyes in four solvents on thin-layer plates coated with cellulose with two standard dyes and then giving the dyes a code depending on where they travel to in relation to the two standard dyes. This code is compared with the list of codes given in Table II thereby giving an initial identification of the dyes. The identity of the food colour is then confirmed by running in solvents together with spots of the suspected food colours.

## TABLE III

CHROMATOGRAPHIC SOLVENTS USED IN THE THIN-LAYER CHROMATOGRAPHIC SEPARATION OF THE DYES
Solvents $\mathbf{1 - r o}$ are used with cellulose plates; solvents $\mathbf{1 r - 1 5}$ are used with silica gel plates.

| Solvent No. | Composition | Reference |
| :---: | :---: | :---: |
| 1 | Trisodium citrate ( 2 g ), water ( 85 ml ), 0.88 ammonia ( 15 ml ) | 1 |
| 2 | tert.-Butanol-propanoic acicl-water ( $50: 12: 38$ ) | I |
| 3 | Trisodium citrate ( 2 g ), hexamine ( 5 g ), water ( 50 ml ), methanol ( 50 ml ) | 3 |
| 4 | 2-Methyl propan-1-ol-water-cthanol-o.88 ammonia (25:25:50:2) | 2 |
| 5 | Propan-r-ol-ethyl acetate-water (6:1:3) | 1 |
| 6 | Butan-r-ol-water-glacial acetic acid (20:12:10) | - |
| 7 | Hydrochloric acid, S.G. I.I8-water ( $23: 77$ ) | - |
| 8 | Butan-r-ol-water-pyridine-ethanol (4:4:2:2) | - |
| 9 | Ethyl methyl ketone-acetone-water-0.83 ammonia ( $70: 30: 30: 0.5$ ) | - |
| ro | 13utan-r-ol-water-ethanol-quinolinc (4:4:3:2) | - |
| II | Propan-2-ol-0.88 ammonia (4:1) | 2 |
| 12 | Propan-2-01-0.88 ammonia (85:15) | - |
| 13 | Methanol-chloroform-water-quinoline (4:2:2:2) | - |
| 14 | Methanol-chloroform-quinoline (4:4:2) | - |
| I 5 | Propan-2-ol-chloroform-water-diethylamine (50:25:20:15) | - |

MATERIALS AND METHODS

## Apparatus

Thin-layer chromatographic apparatus for the preparation of thin layers 0.25 mm thick on $200 \times 200 \mathrm{~mm}$ glass plates. Chromatographic development tanks. $5 \mu \mathrm{l}$ pipettes e.g. Microcap disposable pipettes.

## Reagents

Cellulose powder. Microcrystalline cellulose, available from Applied Science Laboratories Inc. Prepare plates as follows: Shake 20 g cellulose powder with 60 ml methanol for 3 min and blend at high speed for 30 sec . Spread onto plates and air-dry or dry in an oven at $80^{\circ}$.

Silica Gel G. Available from E. Merck. Prepare plates as follows: Shake 30 g Silica Gel G with 60 ml water for I to 2 min . Spread onto plates and, after the layer has set, activate the plates by heating to $105^{\circ}$ for I h.

Reference dye solutions. o.I \% in water.
Chromatographic solvents. See Table III. All solvent mixtures should be freshly prepared.

## Procedure

Place two spots of $1-2 \mu \mathrm{l}$ of the dye solution onto each of four cellulose plates at a distance of at least 20 mm from the edge and bottom of the plate. Also spot on the plates $I-2 \mu$ of a solution of Orange $G$ and a solution of Amaranth as reference spots and place a spot of a mixture of Orange $G$ and Amaranth on top of one of the sample spots. Dry the spots by placing the plates in an oven at $105^{\circ}$ for $5-10$ min. Develop the cooled plates in solvents $1,2,3$ and 4 for a length of run of about 150 mm at room temperature. Remove the plates from the tanks and allow them to air dry. When the plates are dry rule lines across so as to divide the plates into the following sections: code A: spots travelling above Orange G; code B: spots travelling with Orange G; code C: spots travelling below Orange G but above Amaranth; code D: spots travelling with Amaranth; code E: spots travelling below Amaranth.

Check whether the sample has affected the development characteristics of Orange G and Amaranth and if so make allowance for this when dividing the plate into sections. Observe which section the spots from the sample solution appear in for each plate and write down all possible composite codes for each spot by listing the code individual letters in the order-solvent 1 , solvent 2 , solvent 3 , solvent 4 . Compare the codes with the list given in Table II and hence obtain a preliminary identification of the dyes. When two or more spots are similar in colour, cross code the dyes so that all possible dyes are obtained from Table II. Also if a dye is visible in one solvent but not in another then this indicates that the dye is masked by another dye and so all codes for spots in that solvent must be used in constructing the composite codes. A further identification of the dyes may be obtained by calculating the $\boldsymbol{R}_{\boldsymbol{F}}$ and $\boldsymbol{R}_{\boldsymbol{X}}$ (with respect to Orange G) values and referring to the Tables IV-VII. This will eliminate some of the dyes obtained from Table II. All $R_{F}$ and $R_{X}$ values have been calculated by measuring to the leading edge of the spots.

The identification of the sample dye is then confirmed by chromatography on a plate with standard spots of the suspected colours using suitable solvents. Spots of
J. Chromatog., 54 (1971) 393-404

## CABLE IV

$?_{f}$ and $R_{x}$ (with respect to Orange G) values for red dyes

| Solour | Colour <br> index <br> No. | Approximate $R_{p}$ values |  |  |  |  |  |  |  |  | Approximate $R_{\boldsymbol{x}}$ values |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | Solvent No. |  |  |  |  |  |  |  |  | Solvent No. |  |  |  |  |  |  |  |  |
|  |  | $I$ | 2 | 3 | 4 | 5 | 6 | 8 | 9 | II | $I$ | 2 | 3 | 4 | 5 | 6 | 8 | 9 | rI |
| Amaranth | 16185 | 0.6 | 0.3 | 0.5 | 0.6 | 0.4 | 0.2 | 0.6 | 0.4 | 0.4 | 0.8 | 0.4 | 0.5 | 0.8 | 0.7 | 0.4 | 0.8 | 0.4 | 0.9 |
| Bordeaux B | 16180 | 0.2 | 0.6 | 0.4 | 0.6 | 0.5 | 0.6 | 0.7 | 1.0 | 0.4 | 0.3 | 0.9 | 0.4 | 0.8 | 0.9 | I. 0 | 1.0 | 1.0 | 0.9 |
| Carmoisine | 14720 | 0.3 | 0.7 | 0.6 | 0.5 | 0.7 | 0.6 | 0.8 | 0.9 | 0.4 | 0.4 | I. 1 | 0.6 | 0.7 | 1.1 | 1.0 | I. 1 | 0.9 | 0.9 |
| Eosine | 45380 | 2 | 1.0 | 0.7 | 0.8 | I. ${ }^{\text {a }}$ | 1.0 | 0.9 | 1.0 | 0.6 | 0.3 | 1.5 | 0.7 | I.I | 1.6 | I. 7 | 1.3 | 1.0 | I. 4 |
| Erythrosine | 45430 | 0.1 | 1.0 | 0.7 | 0.9 | 1.0 | 1.0 | 0.9 | I. 0 | 0.7 | 0.2 | 1. 5 | 0.7 | I. 2 | 1.6 | 1.7 | I. 3 | 1.0 | I. 6 |
| Fast Red E | 16045 | 0.4 | 0.7 | 0.6 | 0.7 | 0.6 | 0.5 | 0.8 | 1.0 | 0.4 | 0.6 | I. 0 | 0.6 | I. 0 | 1.0 | 0.9 | I.I | 1.0 | 0.9 |
| Ponceal 3 R | 16155 | 0.2 | 0.6 | 0.4 | 0.6 | 0.5 | 0.5 | 0.8 | 0.9 | 0.3 | 0.3 | 0.9 | 0.4 | 0.8 | 0.7 | 0.9 | 1.1 | 0.9 | 0.7 |
| Ponceau 4R | 16255 | 0.7 | 0.5 | 0.9 | 0.6 | 0.4 | 0.3 | 0.7 | 0.6 | 0.2 | r. 0 | 0.6 | ז. 0 | 0.8 | 0.7 | 0.5 | 1.0 | 0.6 | 0.4 |
| Ponceau 6R | 16290 | 0.8 | 0.2 | 0.8 | 0.4 | 0.3 | 0.1 | 0.6 | 0.2 | -. 1 | I.I | 0.2 | 0.8 | 0.5 | 0.4 | 0.2 | 0.8 | 0.2 | 0.2 |
| Ponceau MX | 16150 | 0.2 | 0.7 | 0.5 | 0.6 | 0.5 | 0.5 | 0.8 | 0.9 | 0.4 | 0.3 | 0.9 | 0.5 | 0.8 | 0.9 | 0.9 | I.I | 0.9 | 0.8 |
| Ponceau SX | 14700 | 0.4 | 0.7 | 0.5 | 0.6 | 0.5 | 0.5 | 0.8 | 0.9 | 0.4 | 0.6 | 0.9 | 0.5 | 0.8 | 0.9 | 0.9 | I. 1 | 0.9 | 0.8 |
| Red 2G | 18050 | - 0.6 | 0.6 | 0.7 | 0.6 | 0.4 | 0.5 | 0.7 | 0.9 | 0.4 | 0.8 | 0.8 | 0.7 | 0.8 | 0.7 | 0.9 | 1.0 | 0.9 | 0.9 |
| Red 6B | 18055 | 0.4 | 0.3 | 0.4 | 0.5 | 0.4 | 0.2 | 0.6 | 0.5 | 0.4 | 0.6 | 0.4 | 0.4 | 0.7 | 0.7 | 0.4 | 0.8 | 0.5 | 0.9 |
| Red rob | 17200 | 0.2 | 0.5 | 0.6 | 0.6 | 0.4 | 0.3 | 0.7 | 0.8 | 0.4 | 0.3 | 0.6 | 0.6 | 0.8 | 0.7 | 0.5 | 1.0 | 0.8 | 0.9 |
| Red FB | 14780 | 0.0 | 0.3 | 0.1 | 0.2 | 0.4 | 0.2 | 0.7 | 0.4 | 0.6 | 0.0 | 0.4 | -. 1 | 0.3 | 0.7 | 0.4 | 1.0 | 0.4 | 1.3 |
| Rhodamine B | 45170 | 0.5 | 1.0 | 0.9 | 1.0 | 1.0 | 1.0 | 0.9 | I. 0 | 0.8 | 0.7 | 1.5 | I. 0 | I. 3 | 1.6 | 1.7 | I. 4 | I. 0 | 1.8 |
| Scarlet GN | 14815 | 0.9 | 0.7 | 0.9 | 0.8 | 0.8 | 0.6 | 0.8 | I. 0 | 0.5 | I.I | 0.9 | 1.0 | I.I | 1.2 | 1.0 | I. | I. 0 | 1.2 |

TABLE V
$R_{F}$ and $R_{X}$ (with respect to Orange G) values for yellow and orange dyes

| Colour | Colour index No. | Approximate $R_{\text {F }}$ values |  |  |  |  |  |  |  |  | Approximate Rx values |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | Solvent No. |  |  |  |  |  |  |  |  | Solvent No. |  |  |  |  |  |  |  |  |
|  |  | $\boldsymbol{r}$ | 2 | 3 | 4 | 5 | 6 | 7 | 8 | II | $I$ | 2 | 3 | 4 | 5 | 6 | 7 | 8 | II |
| Auramine | 41000 | 0.3 | I. 0 | 0.9 | 1.0 | 0.9 | 1.0 | not <br> vis- <br> ible | 0.8 | 0.8 | 0.4 | r. 4 | 1.0 | 1.6 | I. 4 | I. 9 | not <br> vis- <br> ible | I. 3 | x. 8 |
| Acid Yellow | 13015 | . 0.7 | 0.6 | 0.9 | 0.6 | 0.6 | 0.5 | 0.7 | 0.7 | 0.4 | 0.8 | 0.9 | 1.0 | 0.9 | 0.9 | 0.9 | 1.1 | 1.0 | 1.0 |
| Chrysoidine | 11270 | - 0.1 | 0.8 | 0.7 | 0.9 | 0.8 | 0.9 | 0.1 | 0.9 | 0.8 | 0.2 | 1.2 | 0.7 | 1.6 | 1.2 | r. 7 | 0.2 | 1.4 | r. 8 |
| Chrysoin S | 14270 | 0.5 | 0.8 | 0.8 | 0.6 | 0.9 | 0.7 | 0.4 | 0.8 | 0.5 | 0.6 | 1.2 | 0.9 | 0.9 | 1.3 | 1.4 | 0.7 | 1.2 | 1.0 |
| Naphthol Yellow S | 10316 | 0.6 | 0.7 | 0.8 | 0.7 | 0.7 | 0.6 | not <br> vis- <br> ible | 0.7 | 0.5 | 0.7 | I. | 0.9 | I.I | I.I | I.I | not visible | 1.I | I. 0 |
| Orange G | 16230 | 0.8 | 0.7 | 0.9 | 0.6 | 0.7 | 0.5 | 0.7 | 0.7 | 0.4 | 1.0 | I. 0 | 1.0 | 1.0 | 1.O | 1.0 | 1.0 | 1.0 | 1.0 |
| Orange GGN | 15980 | 0.6 | 0.7 | 0.8 | 0.6 | 0.6 | 0.5 | 0.2 | 0.7 | 0.4 | 0.8 | I. 0 | 0.9 | 1.0 | I. 0 | 1.0 | 0.3 | 1.0 | 1.0 |
| Orange I | 14600 | 0.4 | 0.8 | 0.8 | 0.7 | 0.9 | 0.7 | 0.1 | 0.8 | 0.5 | 0.6 | 1.2 | 0.9 | I.I | 1.3 | I. 4 | 0.2 | 1.2 | 1.2 |
| Orange RN | 15970 | 0.4 , | 0.9 | 0.7 | 0.6, | 0.6, | 0.5, | 0.1 | 0.7 , | 0.7 | 0.5 | 1. | 0.8 | 0.9, | o.8, | 0.9, | . 2 | 1.0, | 1.6 |
| Quinoline | 47005 | 0.5 0.5 |  | 0.4 | 0.8 0.6 | 0.9 0.7 | 0.8 0.6 | 0.1 | 0.8 | 0.7 | 0.7 0.1 0.1 | I. 0 | 0.5 | 1.3 10 | I. I I | 1.5 1.2 | -. 1 | 1.2 0.9 | 1. 6 |
| Yellow | 47005 | 0.3 |  |  |  |  |  |  | 0.7 |  | 0.4 |  |  |  |  |  |  | 1.0 |  |
| Sunset Yellow | 15985 | 0.6 | 0.7 | 0.8 | 0.6 | 0.6 | 0.5 | 0.2 | 0.7 | 0.4 | 0.7 | 0.9 | 0.9 | 10 | 0.9 | 1.0 | 0.4 | 1.0 | 1.O |
| Tartrazine | 19140 | 0.8 | 0.4 | 0.8 | 0.4 | 0.5 | 0.3 | 0.4 | 0.5 | 0.3 | 1.0 | 0.6 | 0.8 | 0.6 | 0.7 | 0.6 | 0.6 | 0.7 | 0.6 |
| Yellow 2G | r8965 | 0.9 | 0.8 | 1.0 | 0.6 | 0.8 | 0.6 | 0.9 | 0.7 | 0.4 | I.I | 1.2 | 1.I | r.o | 1.2 | 1.2 | 1.5 | 1.0 | I. 0 |

TABLE VI
$R_{F}$ and $\boldsymbol{R}_{\boldsymbol{X}}$ (with respect to Orange $\dot{G}$ ) values for brown and violet dyes

| Colour | Colour Approximate $R_{F}$ values index <br> No. Solvent No. |  |  |  |  |  |  |  |  | Approximate $\boldsymbol{R}_{\boldsymbol{X}}$ values |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  |  |  |  |  |  |  | Solvent No. |  |  |  |  |  |  |
|  |  | I | 2 | 3 | 4 | 5 |  | 14 | 15 | $I$ | 2 | 3 | 4 | 5 | II 14 | 15 |
| Brown FK | - | streak | streak | streak | 0.6 | streak |  | streak | $\begin{aligned} & 0.4 \\ & 0.5 \end{aligned}$ | streak | streak | streak | 0.9 | streak | I.I, streak 1.3 | 0.6, $0.7$ |
| Chocolate Brown FB | - | streak | streak | streak | streak | streak |  |  | small streak | streak | streak | streak | streak | streak | 0.00 .0 | small <br> streak |
| Chocolate Brown HT | 20285 | streak | streak | streak | streak | streak |  |  | long streak | streak | streak | streak | streak | streak | 0.00 .0 | long streak |
| Acid Magenta | 42685 |  | $\begin{aligned} & 0.4 \\ & 0.6 \end{aligned}$ | 0.9 | 0.6 | $\begin{aligned} & 0.5, \\ & 0.6 \end{aligned}$ |  | not visible | not visible | I. 4 | $\begin{aligned} & 0.7, \\ & \text { I. } 0 \end{aligned}$ | 1.I | 0.8 | $\begin{aligned} & 0.8, \\ & 0.9 \end{aligned}$ | 0.3 not visible | not visible |
| Methyl Violet | 42535 | streak | I. 0 | $\begin{aligned} & 0.8, \\ & 0.9 \end{aligned}$ | 1.0 | r.0 | $\begin{aligned} & 0.8, \\ & 0.9 \end{aligned}$ | $0.8$ | 0.7 | streak | 1.5 | I.O, | I. 4 | 1. 6 | $\begin{aligned} & \text { I.3, } 1.9 \\ & \text { I. } 4 \end{aligned}$ | I. 1 |
| Violet BNP | - | 0.7 | 0.8 | 0.9 | 0.9 | 0.9 | 0.5 | 0.4 | 0.5 | 1.0 | 1.3 | 1.1 | 1.3 | 1.4 | 0.7 1.1 | 0.7 |
| Violet 5BN | 42650 |  | 0.8 | 0.9 | 0.9 | 0.9 | 0.5 | 0.4 | 0.5 | 1.0 | 1.3 | I.I | I. 3 | 1.4 | 0.7 I.I | 0.7 |
| Violet 6B | 42640 |  | 0.8 | 1.0 | 1.0 | 0.9 | $\begin{aligned} & 0.5, \\ & 0.6, \\ & 0.7 \end{aligned}$ | $\begin{aligned} & 0.4, \\ & 0.5 \end{aligned}$ | 0.6 | 0.8 | 1.3 | 1.2 | I. 4 | 1.4 | $\begin{aligned} & 0.8, \text { I.I, } \\ & 0.9, \text { I. } 2 \\ & \text { I.0 } \end{aligned}$ | 0.9 |

## ABLE VII

fand $R_{X}$ (with respect to Orange G) values for green, blue and black dyes

| slour | Colour <br> index <br> No. | Approximate $R_{F}$ values |  |  |  |  |  |  |  |  | Approximate $R_{\boldsymbol{X}}$ values |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | Solvent No. |  |  |  |  |  |  |  |  | Solvent No. |  |  |  |  |  |  |  |  |
|  |  | $I$ | 2 | 3 | 4 | 5 | II | 12 | 13 | 10 | $I$ | 2 | 3 | 4 | 5 | $\boldsymbol{r}$ I | 12 | 13 | Io |
| ast Green FCF reen S uinea Green B ight Green | 42053 | 0.9 | 0.8 | 1.0 | 0.8 | 0.8 | 0.1 | 0.1 | 0.8 | 0.8 | 1.2 | I.I | I.I | 1.0 | 1.1 | 0.5 | 0.2 | 1.0 | I. 0 |
|  | 44090 | 0.9 | 0.8 | 0.9 | 0.8 | 0.8 | 0.1 | 0.1 | 0.7 | 0.8 | 1.2 | 1.1 | 1.0 | 1.0 | 1.2 | 0.4 | 0.2 | 0.9 | 1.0 |
|  | 42085 | 0.7 | 0.9 | 1.0 | 1.0 | 0.9 | 0.3 | 0.3 | 0.8 | 0.9 | 0.9 | 1.2 | I. 1 | 1.3 | 1.3 | I.I | I. 4 | 1.0 | 1.2 |
|  | 42095 | 0.9 | 0.8 | 1.0 | 0.9 | 0.8 | 0.3 | o. 1 | 0.8 | 0.8 | I.I | 1.1 | I. 1 | I. 2 | I.I | 0.9 | 0.5 | I. 0 | . 0 |
| lue VRS | 42045 |  | 0.9 | I. 0 | 0.9 | 0.9 | 0.3 | 0.2 | 0.8 | 0.9 | I. 1 | 1.2 | r.I | 1.2 | 1.3 | I. 1 | 1.0 | 1.0 | I. 2 |
|  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| adanthrene |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| Blue ${ }^{\text {a }}$ | 69800 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 |
| adigo Carmine | 73015 | 0.2, | 0.4 | 0.5 | 0.6 | 0.5 | 0.3 | 0.2 | 0.8 | 0.7 | 0.3 | 0.6 | 0.6 | 0.8 | 0.7 | 1.0 | 1.0 | 1.0 | 0.9 |
|  |  | 0.3 |  |  |  |  |  |  |  |  | 0.4 |  |  |  |  |  |  |  |  |
| atent Blue V | 42051 | 0.9 | 0.9 | 1.0 | 0.9 | 0.9 | O. 1 | 0.0 | 0.8 | 0.9 | 1.2 | 1.2 | 1.1 | 1.1 | 1.3 | 0.2 | 0.0 | 1.0 | 1.2 |
| lack 7984 | 27755 | 0.2 | 0.3 | 0.2 | 0.4 | 0.4 | O.r | 0.0 | 0.7 | 0.5 | 0.2 | 0.4 | 0.2 | 0.6 | 0.5 | 0.4 | 0.0 | 0.9 | 0.7 |
| lack PN | 28440 | 0.4 | 0.3 | 0.2 | 0.4 | 0.4 | O. 1 | 0.0 | 0.7 | 0.7 | 0.4 | 0.4 | 0.2 | 0.6 | 0.5 | 0.4 | 0.0 | 0.9 | 0.9 |

a Indanthrene Blue is insoluble in water and most organic solvents.
the sample solution are also overspotted with spots of the suspected dyes. The unknown dye is identified by giving a single spot with the correct standard while all the other standards give rise to double spots.

If the sample contains several dyes, more than one solvent may be necessary for complete confirmation of the dyes.

## DISCUSSION

In constructing the table of codes for the dyes, slight variations in the development characteristics of the dyes have been taken into account so that some dyes occur under a number of different codes. Brown FK, Chocolate Brown FB and Chocolate Brown HT have not been included in this table as they streak in the solvents used. If the standard dyes, Orange $G$ and Amaranth, run very differently when overspotted on the sample from when they are spotted separately on the plate then the spots in the sample should be coded twice, once using the standards in the sample to divide up the plate and once using the standard spotted separately to divide up the plate. By this means all possible dyes will be obtained, but a number of these will be rejected on the basis of colour and $R_{F}$ value. However, do not discount dyes which could give rise to the colour of the spot, e.g. an orange coloured spot may be a red and yellow dye superimposed.

As most problems arise from the possibility of a red and yellow dye being together in the mixture, the separation of the reds, oranges and yellows are set out in Table VIII. All $\boldsymbol{R}_{\boldsymbol{F}}$ and $\boldsymbol{R}_{\boldsymbol{X}}$ values have been calculated by measuring to the leading edge of a spot as this was found to be more reliable for spots which tail. When confirming the identity of a dye by running it with standard dyes it is useful to observe

TABLE VIII.
SEPARATION OF REDS, ORANGES AND YELLOWS IN SOLVENTS I, 2,3 AND 4
Dyes in italics are completely separated from the others.


$$
R_{F}=0
$$

the plate under UV light of 254 nm and 350 nm as some of the dyes fluoresce. The following mixtures of dyes could not be separated in any of the solvents tried: Chocolate Brown HT and Chocolate Brown FB, Ponceau 3R and Ponceau MX, Violet 5BN and Violet BNP.

Chocolate Brown HT can be tentatively distinguished from Chocolate Brown FB by running in solvent 15 on silica gel. Chocolate Brown FB produces a small streak from the spotting line whereas Chocolate Brown HT produces two spots and a streak from the spotting line. The two spots travel higher than the streak from Chocolate Brown FB.

Aldred ${ }^{6}$ has reported that Violet 5BN and Violet BNP can be separated on silica gel using a mixture of 2 -methyl propan-I-ol, ethanol and water as developing solvent. When this system was tried we did not obtain a separation of the samples of Violet 5BN and Violet BNP which we were using. Some of these dyes may be broken down during extraction from the foodstuffs, or in the foodstuff itself, and the decomposition products may affect the separation of the dyes. Work is in progress on these aspects to see how they will affect the identification scheme and further publications of the results of this work will follow.

Violet 5 BN is permitted only in South Africa and Violet BNP is permitted only in Denmark, New Zealand and the United Kingdom. Consequently the need to separate these two dyes should not arise very often. However, they can be distinguished by their IR spectra. No work has been carried out on extraction of these dyes from foodstuffs and it is realised that co-extractives may affect the running characteristics of various dyes but by overspotting the sample with the suspected dyes in the final confirmation any irregularities should not affect the identification of the dyestuff.


Fig. I. A simple device for the measurement of $R_{F}$ values. A - B, resistance wire mounted on perspex; C, a developed thin-layer plate; $P, I k \Omega$ potentiometer; $V, 2.5 \mathrm{~V} . \mathrm{f} . \mathrm{s} . \mathrm{d}$. voltmeter; D , contact probe.

## THIN-LAYER CHROMATOGRAPHIC TECHNIQUES

## $R_{F}$ measurement

To relieve the tedium of measuring a large number of $R_{F}$ values a simple electrical device was constructed. The device consists of a perspex template which slides over the thin-layer plate. The template has a length of resistance wire stretched between two terminals and a sliding contact for making contact with the resistance wire. The resistancerwire is made part of a simple potentiometer circuit as shown in Fig. I. The template is placed over the thin-layer plate and adjusted so that terminal " $A$ " is over the spotting line. The sliding contact is moved to the solvent front or the standard spot, if $R_{X}$ values are required, and the potentiometer " $P$ " adjusted so that the voltmeter reads 1.0 units. The sliding contact is then moved over the spot whose $R_{F}$ or $R_{X}$ value is required and the voltmeter reading noted. The template is then moved along keeping it in contact with the bottom edge of the plate until it is over the next spot.

## Documentation of chromatograms

Copies of the thin-layer chromatograms were made by a simple blue print type procedure. The spots on the plates are scribed round with a needle and then placed coated surface down on a piece of "Elackline" paper. (Blackline paper for ammonia development, $Z Y_{5} M$, obtainable from Mason Ltd., Colchester.) The back of the plate is illuminated by means of photoflood bulbs for approximately 40 sec . The plate is removed and the paper is suspended in a tank containing a few millilitres of 0.88 ammonia solution for about I min. The print obtained consists of black spots ringed with a white line.

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